

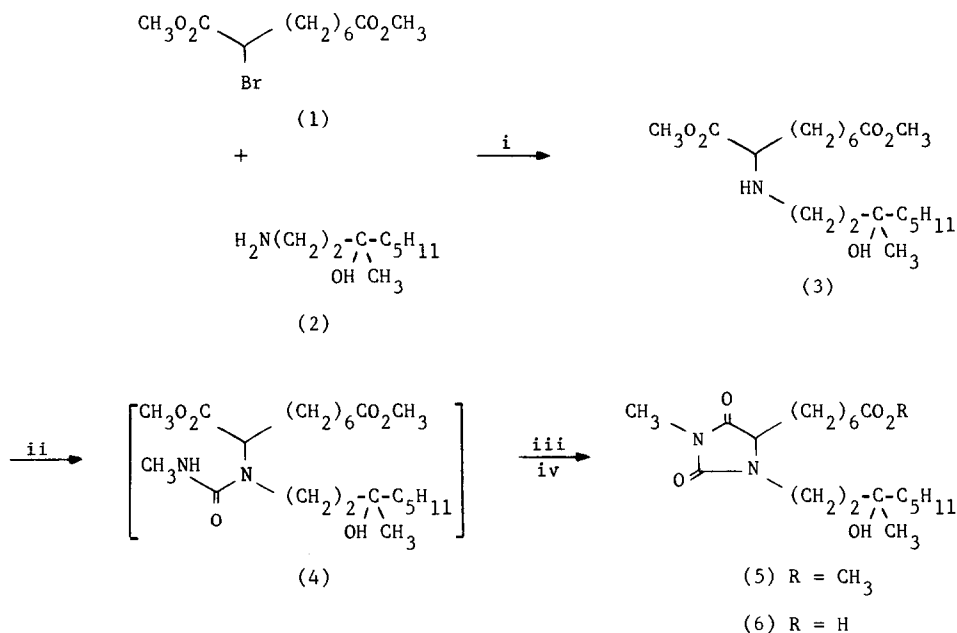
10,12-DIAZAPROSTAGLANDIN ANALOGUES

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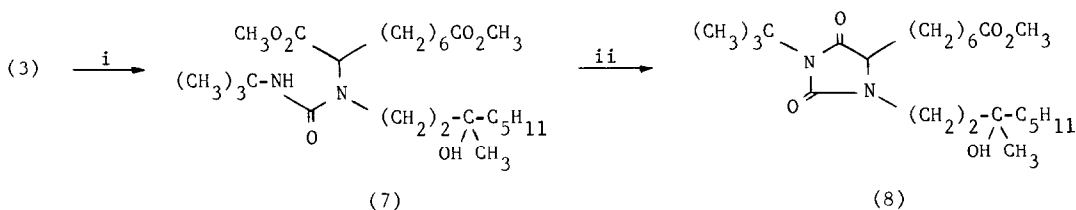
Considerable interest has been shown recently in the synthesis and biological activities of azaprostaglandin analogues.¹⁻⁹ As a continuation of our work in this area,^{3,9} we have developed a versatile, high yielding route (Scheme 1) to 10,12-diazaprostaglandin analogues from readily available starting materials.

Scheme 1



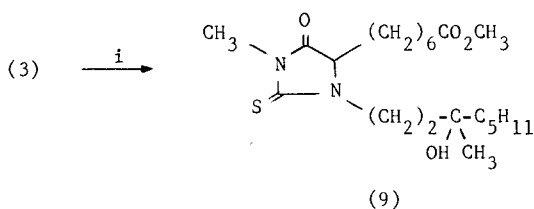
- i; reflux in CH₃OH with Na₂CO₃
ii; CH₃NCO, toluene, room temperature
iii; reflux in toluene, 3 hr.
iv; K₂CO₃, H₂O, CH₃OH, reflux.

Methods for the synthesis of (1), (2) and (3) have been reported elsewhere.³ Reaction of the amino di-ester (3) with methyl isocyanate gave, directly, the 10,12-diazaprostaglandin analogue (5), via thermal cyclisation of the intermediate urea derivative (4). The analogue (5), which was obtained as a colourless gum after column chromatography,¹⁰ was subjected to alkaline hydrolysis to obtain the acid (6). The thermal cyclisation did not occur when tert-butylisocyanate was used in place of methyl isocyanate and the isolated product was the urea (7). However, the required 10,12-diazaprostaglandin analogue (8) was obtained, albeit in low yield, via cyclisation of (7) with potassium tert-butoxide in toluene. The yield of this cyclisation reaction was increased considerably when the diethyl rather than the dimethyl ester was used.



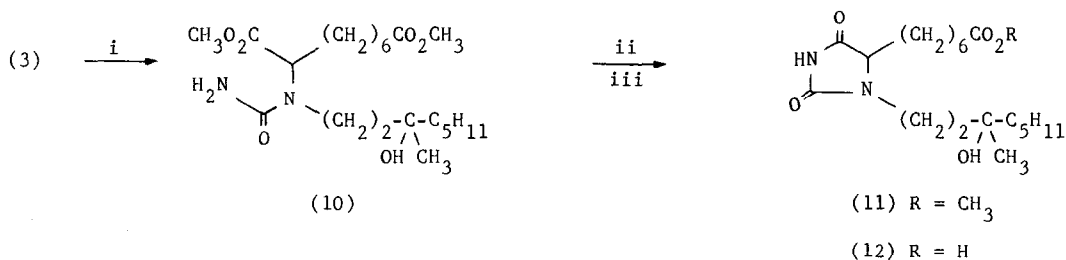
i; $(\text{CH}_3)_3\text{CNCO}$, reflux in toluene, 3 hrs.
 ii; $\text{KOC}(\text{CH}_3)_3$, reflux in toluene, 3 hrs.

The route outlined in Scheme 1 was also used to produce a series of 11-thio-10,12-diazaprostaglandin analogues. For example, reaction of the amino di-ester (3) with methyl isothiocyanate gave the ester (9) as a pale yellow gum after column chromatography.



i; CH_3NCS , reflux in toluene, 3 hrs.

In order to obtain 10-unsubstituted 10,12-diazaprostaglandin analogues, a different approach was necessary. For example, reaction of the amino di-ester (3) with potassium cyanate and dilute hydrochloric acid gave the urea (10) which was thermally cyclised to the 10,12-diazaprostaglandin analogue (11). Subsequent alkaline hydrolysis gave the acid (12).

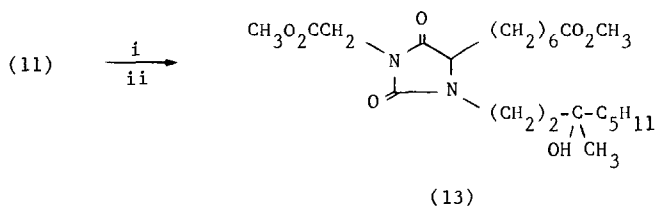


i; 2N HCl, KCNO, 0°C, 0.5 hr.

ii; 100°C, 4 hr.

iii; K₂CO₃, H₂O, CH₃OH, reflux.

The acidic nature of the proton at the 10-position of (11) was utilised in an alternative route to the 10-substituted analogues. Thus, conversion of (11) to its 10-sodium salt, followed by treatment with methyl bromoacetate gave the di-ester (13) as a colourless gum after column chromatography.



i; NaH, DMF, room temperature, 2 hrs.

ii; CH₃O₂CCH₂Br, DMF, room temperature, 5 hrs.

The 10,12-diazaprostaglandin analogues were obtained as mixtures of diastereoisomers.

Data for compounds (5) to (13)

Compound (5): Yield 82% from (3). IR: ν_{max} , (film), 3450, 1765, 1735, 1705 cm⁻¹.

NMR¹¹: τ , (CDCl₃), 7.75 (m, 2H, CH₂CO₂CH₃); 7.25 (s, 1H, OH); 7.07 (s, 3H, NCH₃); 6.7 (brm, 2H, NCH₂); 6.38 (s, 3H, CO₂CH₃); 6.05 ppm (m, 1H, NCH). Mass spectrum: m/e (m*) requires 398.2778; found 398.2757.

Compound (6): Yield 86% from (5). IR: ν_{max} , (film), 3600 to 2400, 1760, 1700 (broad) cm⁻¹

NMR: τ , (CDCl₃), 7.65 (t, 2H, CH₂CO₂H); 7.0 (s, 3H, NCH₃); 6.6 (brm, 2H, NCH₂); 5.95 (t, 1H, NCH); 3.65 ppm (bs, 2H, CO₂H, OH). Mass spectrum: m/e (m*) requires 384.2622; found 384.2601.

Compound (7): Yield 85% from (3). IR: ν_{max} , (film), 3350, 1740, 1630 cm⁻¹. NMR: τ , (CCl₄), 7.75 (t, 2H, CH₂CO₂CH₃); 7.1 (bs, 1H, OH); 6.8 (m, 2H, NCH₂); 6.4 (s, 3H, CO₂CH₃); 6.35 ppm (s, 3H, CO₂CH₃). Mass spectrum: m/e (m*) requires 472.3512; found 472.3511.

Compound (8): Yield 15% from (7). IR: ν_{\max} , (film), 3450, 1760, 1740, 1700 cm^{-1} . NMR: τ , (CCl_4), 7.75 (t, 2H, $\text{CH}_2\text{CO}_2\text{CH}_3$); 7.1 (s, 1H, OH); 6.8 (brm, 2H, NCH_2); 6.4 (s, 3H, CO_2CH_3); 6.25 ppm (m, 1H, NCH). Mass spectrum: m/e (m^*) requires 440.3331; found 440.3258.

Compound (9): Yield 88% from (3). IR: ν_{\max} , (film), 3470, 1740 cm^{-1} . NMR: τ , (CCl_4), 7.85 (t, 2H, $\text{CH}_2\text{CO}_2\text{CH}_3$); 7.75 (s, 1H, OH); 6.7 (brm, 2H, NCH_2); 6.82 (s, 3H, NCH_3); 6.4 (s, 3H, CO_2CH_3); 5.9 ppm (brm, 1H, NCH). Mass spectrum: m/e (m^*) requires 414.2549; found 414.2565.

Compound (10): Yield 100% (crude) from (3). IR: ν_{\max} , (film), 3600 to 2800, 1720, 1660, 1600 cm^{-1} . NMR: τ , (CDCl_3), 7.75 (t, 2H, $\text{CH}_2\text{CO}_2\text{CH}_3$); 6.8 (brm, 2H, NCH_2); 6.4 (s, 3H, CO_2CH_3); 5.95 ppm (m, 1H, NCH).

Compound (11): Quantitative yield from (10). IR: ν_{\max} , (film), 3450, 3200, 1760, 1710 cm^{-1} . NMR: τ , (CCl_4), 7.75 (t, 2H, $\text{CH}_2\text{CO}_2\text{CH}_3$); 6.7 (brm, 2H, NCH_2); 6.35 (s, 3H, CO_2CH_3); 5.95 ppm (m, 1H, NCH). Mass spectrum: m/e (m^*) requires 384.2622; found 384.2620.

Compound (12): Quantitative yield from (11). IR: ν_{\max} , (film), 3600 to 2500, 3400, 3200, 1760, 1710 cm^{-1} . NMR: τ , (CDCl_3), 7.65 (m, 2H, $\text{CH}_2\text{CO}_2\text{H}$); 6.4 (brm, 2H, NCH_2); 5.9 (bs, 1H, NCH); 3.2 (bs, 2H, CO_2H , OH); 0.55 ppm (bs, 1H, CONHCO). Mass spectrum: m/e (m^*) requires 370.2466; found 370.2473.

Compound (13): Yield 98% from (11). IR: ν_{\max} , (film), 3450, 1760, 1720 (shoulder), 1710 cm^{-1} . NMR: τ , (CCl_4), 7.75 (t, 2H, $\text{CH}_2\text{CO}_2\text{CH}_3$); 7.15 (d, 1H, OH); 6.8 (brm, 2H, NCH_2); 6.4 (s, 3H, CO_2CH_3); 6.25 (s, 3H, CO_2CH_3); 5.95 (m, 1H, NCH); 5.85 ppm (s, 2H, $\text{NCH}_2\text{CO}_2\text{CH}_3$). Mass spectrum: m/e ($m^*-\text{H}_2\text{O}$) requires 438.2726; found 438.2719.

References, Notes and Acknowledgements

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10. Column chromatography was carried out on silica gel using chloroform and/or chloroform methanol mixtures as eluants.
11. NMR spectra were determined on a Perkin Elmer R12A (60 MHz) instrument using TMS as internal standard.

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